

Chapter 12

SECTION 12

Problem: Answer the following questions in the space provided.

1. Illustration for the pressure of a fixed mass of gas with constant volume, at two different in the following conditions:

- (a) Increase: temperature increases, volume kept constant
- (b) Decrease: temperature decreases, volume kept the same
- (c) Increase: temperature decreases, volume kept the same
- (d) Decrease: volume decreases, temperature kept the same

2. Two sealed flasks, A and B, contain the different gases at equal volume at the same temperature. Assume that flask A is connected to flask B by a tube. With the pressure in the two flasks remains equal? If not, which flask will have the higher pressure? Will the pressure will not increase equal. If all other factors remain constant, flask A will have higher pressure because it is at a higher temperature.

Problem: Answer the following questions in the space provided.

3. A large flask is filled with air at 20°C. Assume that the volume of the gas is constant. Approximately what is the volume?

4. If the volume of the gas is constant, 20°C, by what factor, does the pressure in the gas increase or decrease? Assume the volume of the gas is constant, constant. (Temperature increases as temperature rises)

5. Assume the temperature is constant. How does the pressure change? (Temperature does not change)

6. Assume the pressure is constant. How does the temperature change? (Temperature does not change)

Review Gases Section 2 Short Answer

E. L. Cussler



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