

Table of Elements

Use the table to answer the questions.

Name _____

Element	Symbol	Number of Protons (Atomic #)	Number of Electrons	Number of Neutrons (most stable isotope)	Atomic Mass (most stable isotope)	Type of Element
Hydrogen	H	1	1	0	1	non-metal
Helium	He	2	2	2	4	noble gas
Lithium	Li	3	3	4	7	alkali metal
Carbon	C	6	6	6	12	non-metal
Oxygen	O	8	8	8	16	non-metal
Silicon	Si	14	14	14	28	metalloid
Chlorine	Cl	17	17	18	35	halogen
Calcium	Ca	20	20	20	40	alkaline earth metal
Manganese	Mn	25	25	30	55	transition metal
Iron	Fe	26	26	30	56	transition metal
Lead	Pb	82	82	125	207	post-transition metal
Gold	Au	79	79	118	197	transition metal

- What is the chemical symbols for iron, lead, and gold? _____
- Which element has the fewest protons? _____
- In this table, which element has the most neutrons? _____
- Which element in this table is a noble gas? _____
- Which element has 26 protons? _____
- Which elements have 30 neutrons? _____
- The atomic number (# of protons) uniquely determines an element. Does the number of neutrons determine an element? _____
- A proton has a positive charge; an electron has a negative charge. The number of protons and the number of _____ is the same (this gives the atom a neutral charge).
- How many protons plus neutrons does chlorine have? _____
- The protons and neutrons are located in the nucleus of an atom; tiny electrons orbit around the nucleus. Atomic mass is the weight of the atomic nucleus; it is the number of protons plus the number of _____.

Review Elements And Their Properties Answers Key

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