



Unit: Chemical Bonding Answer Key “Covalent Bonding” - WS #3

Directions: Please answer each question using your notes from today’s class. ☺

1. What is the difference between a molecular formula, structural formula and an electron dot formula? Give an example of each.

A molecular formula gives the chemical make-up of one molecule. It does not show atom-to-atom connectivity. Example: C_3H_8

A structural formula uses lines instead of dots to show shared pairs of electrons.

Example: $H-H$

An electron dot formula uses dots instead of lines to show shared pairs of electrons.

Example: $H:H$

2. What is the difference between single, double, and triple covalent bonds? How many shared electrons are in each?

The difference between the types of bonds is: a single bond is two shared electrons, a double is four shared electrons, and a triple is six shared electrons.

3. Draw the electron dot and structural formulas for the following molecules.

Substance	Rough copy (work!)	Electron dot	Structural
a) Cl_2		<i>Note: The electron dot structures have dots in between each atom instead of lines.</i>	$:\ddot{Cl}-\ddot{Cl}:$
b) H_2S			$\begin{array}{c} :\ddot{S}: \\ \\ H \end{array} - H$
c) BrI			$:\ddot{I}-\ddot{Br}:$

Review Answers For Covalent Bonding

Neil D. Jespersen, Alison Hyslop



Review Answers For Covalent Bonding:

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Chemistry Richard Post, Chad Snyder, Clifford C. Houk, 2020-09-16 A practical complete and easy to use guide for understanding major chemistry concepts and terms Master the fundamentals of chemistry with this fast and easy guide Chemistry is a fundamental science that touches all other sciences including biology physics electronics environmental studies astronomy and more Thousands of students have successfully used the previous editions of Chemistry Concepts and Problems A Self Teaching Guide to learn chemistry either independently as a refresher or in parallel with a college chemistry course This newly revised edition includes updates and additions to improve your success in learning chemistry This book uses an interactive self teaching method including frequent questions and study problems increasing both the speed of learning and retention Monitor your progress with self tests and master chemistry quickly This revised Third Edition provides a fresh step by step approach to learning that requires no prerequisites lets you work at your own pace and reinforces what you learn ensuring lifelong mastery Master the science of basic chemistry with this innovative self paced study guide Teach yourself chemistry refresh your knowledge in preparation for medical studies or other coursework or enhance your college chemistry course Use self study features including review questions and quizzes to ensure that you re really learning the material Prepare for a career in the sciences medicine or engineering with the core content in this user friendly guide Authored by expert postsecondary educators this unique book gently leads students to deeper levels and concepts with practice critical thinking problem solving and self assessment at every stage

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Visualizing Everyday Chemistry Douglas P. Heller, Carl H. Snyder, 2015-01-20 Visualizing Everyday Chemistry Binder Ready Version is for a one semester course dedicated to introducing chemistry to non science students It shows what chemistry is and what it does by integrating words with powerful and compelling visuals and learning aids With this approach students not only learn the basic principles of chemistry but see how chemistry impacts their lives and society The goal of Visualizing Everyday Chemistry Binder Ready Version is to show students that chemistry is important and relevant not because we say it is but because they see it is This text is an unbound binder ready version **Material Science** S. L. Kakani, Amit Kakani, 2004 The Book Has Been Designed To Cover All Relevant Topics In B E Mechanical Metallurgy Material Science Production Engineering M Sc Material Science B Sc Honours M Sc Physics M Sc Chemistry Amie And Diploma Students Students Appearing For Gate Upsc Net Slet And Other Entrance Examinations Will Also Find Book Quite Useful In Nineteen Chapters The Book Deals With Atomic Structure The Structure Of Solids Crystal Defects Chemical Bonding Diffusion In Solids Mechanical Properties And Tests Of Materials Alloys Phase Diagrams And Phase Transformations Heat Treatment Deformation Of Materials Oxidation And Corrosion Electric Magnetic Thermal And Optical Properties Semiconductors Superconductivity Organic Materials Composites And Nanostructured Materials Special Features

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Viselli, 2025-02-25 A bestselling title in this highly regarded review series Lippincott Illustrated Reviews Biochemistry is the go to resource for both faculty and students for mastering the essentials of biochemistry The fully revised 9th Edition helps students quickly review assimilate and integrate large amounts of critical and complex information with unparalleled illustrations that bring concepts to life An intuitive outline organization chapter summaries and review questions that link basic science to real life clinical situations work together to clarify challenging information and strengthen retention and understanding while an emphasis on clinical application updated review tools and accompanying digital resources prepare students for success on course and board exams and beyond

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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully

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