

Fill in the blank in each sentence. DON'T WRITE IN THE MARGINS.

1. The theory that explains the behavior of gases on the molecular level is called the kinetic theory, which is based on assumptions related to theoretical gas behavior called the ideal gas.
2. One the kinetic theory is applied why is helium filled balloons expanding when it is brought inside the space from the cold outside?
 According to the kinetic theory, particles have more kinetic energy. Consequently, the particles in the gas expand outward within the container when the walls are removed. It is called the ideal gas theory. Theoretical relationships for particles within the container are called the ideal gas theory.
3. Define "1 atm. abs."

1 atmosphere is defined as the pressure exerted by a column of mercury 760 mm high at 0°C.

4. Name several ways that total atmospheric pressure can be measured using barometers and using manometers.
5. Pressure is defined as force per unit area. The earth's atmosphere has weight, which creates atmospheric pressure.
6. Standard temperature has been established as 0 °C or 32 °F. Standard pressure has been established as 1.013 atm, 101.3 kPa, 760 mm Hg, or 30 in. Hg. The standard unit for standard temperature and pressure is STP.
7. A barometer is used to measure atmospheric pressure, which is atmospheric pressure or total atmospheric gas pressure.
8. When atmospheric pressure increases, how does the height of the column of mercury change? increases
9. If pressure is constant, the volume of a sample of gas increases as temperature increases.
10. At constant pressure, the volume of a sample of gas is inversely proportional to a temperature relationship with the temperature inversely.

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JL Elias



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