

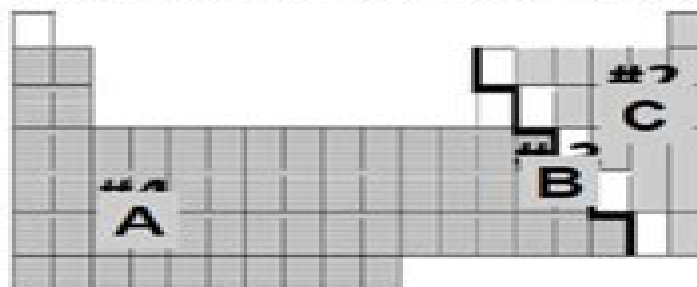
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Review: Unit 4 Periodic Table

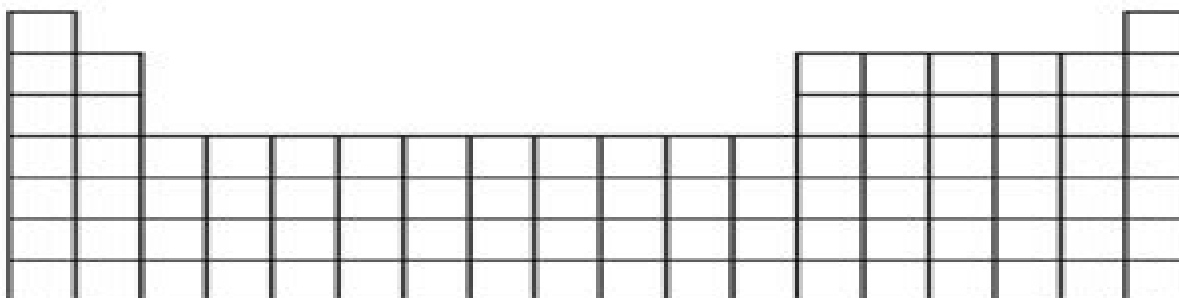
Part 1: History of the Periodic Table/ Families of the PT

1. Dimitri Mendeleev put elements in order based on _____
2. Henry Mosely put elements in order based on _____
3. Label the 3 sections of this periodic table as (Metalloid, Metals, Non-Metals)



Part 2: Periodic Trends

4. Label the periodic table with the following periodic table trends: Atomic Radius, Ionization Energy, Electronegativity (Use different colors for each trend)



5. Where on the Periodic Table is Electronegativity zero?

Part 3: Average Atomic Mass

6. What is the unit for average atomic mass? What does this unit mean, how is it measured?
7. 24.1% of all the isotopes of an element have a mass of 75.23 amu, 48.7% have a mass of 74.61 amu, and 27.2% have a mass of 75.20 amu. What is the average mass of this element?

Review Of The Periodic Table Test Answer

Clifford Lane



Review Of The Periodic Table Test Answer:

Decoding **Review Of The Periodic Table Test Answer**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Review Of The Periodic Table Test Answer**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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