Lab Report: Gravimetric Analysis of an Unknown Sulfate

Experimental Data

Unknown Sulfate ID Code:

1. Mass of empty 250-mL beaker	111.2647
2. Mass of 250-mL beaker and unknown sulfate	112.2273
3. Mass of unknown sulfate	0.3439
4. Mass of empty crucible (without lid)	11.2039
 Mass of crucible (without lid) and barium sulfat 	0 11.6299
6. Mass of barium sulfate	0.4269

Calculations and Conclusions

1. Calculate the mass of sulfate in the unknown sample. Clearly show each step of your

19 Ball 1 - 1001 Ball 1 - 200 (09) - 200 (09) + 200 (09) - 11 00 - 11 00 - 11 0.42 Leg prior x 1mm total x 1 mel total x 1

- 2. Calculate the percentage by mass of sulfate in the unknown sample. Show your work. mass of unbrown sulfate to 3484; mass of sulfate 10.1754 504" 7.504" = 0.1967 x 100 = 51.0204 >.
- 3. Calculate the percentage by mass of the metal in the unknown sample. Show your work.

100.0% - 51.07. = 49.0%.

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RM Cervero

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U.S. Government Research Reports ,1964 English in Analytical Chemistry. Communicating about Methods & Techniques. Книга для студента Надежда Зорина, Александра Соболева, 2022-10-19 Annual Report New York State College of Agriculture and Life Sciences, 1891 Cooperative Chemistry Lab Manual Cooper, 2005-02 The laboratory course described in the lab manual emphasizes experimental design data analysis and problem solving Inherent in the design is the emphasis on communication skills both written and oral Students work in groups on open ended projects in which they are given an initial scenario and then asked to investigate a problem There are no formalized instructions and students must plan and carry out their own investigations Energy Research Abstracts ,1993 Semiannual with semiannual and annual indexes References to all scientific and technical literature coming from DOE its laboratories energy centers and contractors

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