

Experiment 3

Data and Calculations: Resolution of Matter into Pure Substances. II. Fractional Crystallization

Mass of 150-mL beaker 65.24 g

Mass of 50-mL beaker 22.81 g

Mass of ^{SiCl} filter paper 0.57 g

Mass of KNO_3 filter paper 0.51 g

Separation of SiCl

Mass of sample plus 150-mL beaker

86.69 g

Mass of sample $86.69 - 4.28 =$

82.41 g

Mass of SiCl plus filter paper

4.28 g

Mass of SiCl

40.11 g

Percentage of SiCl in sample, by mass

$$\frac{40.11}{86.69} \times 100 = 46\%$$

46 %

~~100~~
40.11

Separation of KNO_3

Mass of 150-mL beaker plus KNO_3

79.06 g

Mass of KNO_3 recovered

13.82 g

$$79.06 - 65.24 = 13.82$$

Percentage (by mass) of sample recovered as KNO_3

17 %

$$\frac{13.82}{79.06} \times 100 =$$

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Resolution Of Matter Into Pure Substances Answers

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