

PHYSICAL SCIENCES P.2

MEMO-2024



basic education

Department:
Basic Education
REPUBLIC OF SOUTH AFRICA

G.12

NATIONAL
SENIOR CERTIFICATE

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

The inorganic reagents shown below are available for reactions IV and Y.

Br ₂	H ₂ SO ₄ (conc.)	NaOH(aq.)	HBr	H ₂
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4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reagent used in reaction IV.



4.8 Write the half-cell reaction for the anode.

4.9 Write the overall cell reaction.



4.10 Write the half-cell reaction for the cathode.

4.11 Write the overall cell reaction.

4.12 The standard electrode potential for the half-cell is +0.34 V.

4.13 Write the standard electrode potential for the half-cell.

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Sebastian Brünink

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Physical Science P2 Memo Grade Exemplar 2012

1. Understanding the eBook Physical Science P2 Memo Grade Exemplar 2012
 - The Rise of Digital Reading Physical Science P2 Memo Grade Exemplar 2012
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science P2 Memo Grade Exemplar 2012
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science P2 Memo Grade Exemplar 2012
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science P2 Memo Grade Exemplar 2012
 - Personalized Recommendations
 - Physical Science P2 Memo Grade Exemplar 2012 User Reviews and Ratings
 - Physical Science P2 Memo Grade Exemplar 2012 and Bestseller Lists
5. Accessing Physical Science P2 Memo Grade Exemplar 2012 Free and Paid eBooks

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 - Physical Science P2 Memo Grade Exemplar 2012 Budget-Friendly Options
6. Navigating Physical Science P2 Memo Grade Exemplar 2012 eBook Formats
- ePub, PDF, MOBI, and More
 - Physical Science P2 Memo Grade Exemplar 2012 Compatibility with Devices
 - Physical Science P2 Memo Grade Exemplar 2012 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Physical Science P2 Memo Grade Exemplar 2012
 - Highlighting and Note-Taking Physical Science P2 Memo Grade Exemplar 2012
 - Interactive Elements Physical Science P2 Memo Grade Exemplar 2012
8. Staying Engaged with Physical Science P2 Memo Grade Exemplar 2012
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science P2 Memo Grade Exemplar 2012
9. Balancing eBooks and Physical Books Physical Science P2 Memo Grade Exemplar 2012
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science P2 Memo Grade Exemplar 2012
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science P2 Memo Grade Exemplar 2012
- Setting Reading Goals Physical Science P2 Memo Grade Exemplar 2012
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science P2 Memo Grade Exemplar 2012
- Fact-Checking eBook Content of Physical Science P2 Memo Grade Exemplar 2012
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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