

Half Life Calculations Practice

2 pages + 2-page Answer Key

HALF-LIFE CALCULATIONS

1. Radioiodine has a half-life of 8 days. If you start with 40 grams of radioiodine, how much will remain after 16 days?
2. If the half-life of Carbon-14 is 5730 years, how much of a 100 gram sample will remain after 11,460 years?
3. If the half-life of Chromium-51 is 28 days, how long will it take for Chromium-51 to decay to 1/8th of the original amount?
4. The half-life of Uranium-238 is 4.5 billion years. How long will it take for a sample of Uranium-238 to decay to 1/8th of the original amount?
5. Bismuth-214 has a half-life of 20 minutes. If you start with 64 grams, how much will remain after 60 minutes?
6. If the half-life of Technetium-99m is 6 hours, how long will it take for Technetium-99m to decay to 1/8th of the original amount?

HALF-LIFE CALCULATIONS

1. Radioiodine has a half-life of 8 days. If you start with 40 grams of radioiodine, how much will remain after 16 days?
 $40 \rightarrow 20 \rightarrow 10$
10g will remain after 16 days
2. If the half-life of Carbon-14 is 5730 years, how much of a 100 gram sample will remain after 11,460 years?
 $100 \rightarrow 50 \rightarrow 25$
25g will remain after 11,460 years
3. If the half-life of Chromium-51 is 28 days, how long will it take for a sample of Chromium-51 to decay to 1/8th of the original amount?
 $1 \rightarrow 1/2 \rightarrow 1/4 \rightarrow 1/8$ 3 half-lives, so 84 days = 3 half-lives
It takes 84 days to decay to 1/8th of the original amount
4. The half-life of Uranium-238 is 4.5 billion years. How long will it take for a sample of Uranium-238 to decay to 1/8th of the original amount?
 $1 \rightarrow 1/2 \rightarrow 1/4 \rightarrow 1/8$ 3 half-lives, so 13.5 billion years = 3 half-lives
It takes 13.5 billion years to decay to 1/8th of the original amount
5. Bismuth-214 has a half-life of 20 minutes. If you start with 64 grams, how much will remain after 60 minutes?
 $64 \rightarrow 32 \rightarrow 16 \rightarrow 8$
8g will remain after 60 minutes
6. If the half-life of Technetium-99m is 6 hours, how long will it take for Technetium-99m to decay to 1/8th of the original amount?
 $1 \rightarrow 1/2 \rightarrow 1/4 \rightarrow 1/8$ 3 half-lives, so 18 hours = 3 half-lives
It takes 18 hours to decay to 1/8th of the original amount

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