

Gay-Lussac's Law

6. The gas in an aerosol can is at a temperature of 27°C and a pressure of 5 atm. If the gas reaches a pressure of 9 atm, what will the can's temperature be?
7. An oxygen tank reads 150 kPa with a temperature of 25°C . What will the pressure read if the tank is cooled to -183°C ?

Combined Gas Law

8. A gas has a volume of 35 liters at STP. What will its volume be at 5.0 atm and 35°C ?
9. 0.4 L of a gas is contained at 129 atm and 273 K. What will the volume of the gas be at 15 atm and 293K?
10. 400 mL of gas is stored at STP. The volume changes to 560 mL, and the temperature to 30°C . What is the pressure (in atm)?

Ideal Gas Law

11. If 6 moles of a gas is at a pressure of 6.6 atm and a volume of 12 liters, what is the temperature of the gas?
12. If there is unknown quantity of gas at a pressure of 132.6 kPa, a volume of 25 liters, and a temperature of 87°C , how many moles of gas are in the sample?
13. If there are 5 moles of gas in a 50-liter container, at a temperature of 250 K, what is the pressure of the gas within the container?
14. If 10.5 moles of gas are stored at a pressure of 0.15 atm and a temperature of 25°C , what is the volume of the container that the gas is stored in?

Dalton's Law

15. A storage container holds a mixture of oxygen, helium, and carbon dioxide. If the partial pressures of the three gases tank are 15 atm of O_2 , 8 atm of He, and 25 atm of CO_2 , what is the total pressure inside of the container?
16. The total pressure of a gas mixture is 1.9 kPa. The partial pressure of gas A is 0.06 kPa and the partial pressure of gas B is 0.25 kPa. What is the partial pressure of the remaining gas in the mixture?
17. If I place 50 moles of N_2 and 1.5 moles of CO_2 in a 40 L container at a temperature of 25°C , what will the pressure of the resulting mixture of gases be?

Science Notebookthe Gas Laws Answer Key

**David I. Ketcheson,Randall J.
LeVeque,Mauricio J. del Razo**



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Educators Grade Guide to Free Teaching Aids ,1963 Redefining Teacher Education and Teacher Preparation Programs in the Post-COVID-19 Era Bull, Prince Hycy,Patterson, Gerrelyn Chunn,2021-12-17 Due to the COVID 19 pandemic teacher preparation programs modified their practices to fit the delivery modes of school districts while developing new ways to prepare candidates Governmental agencies established new guidelines to fit the drastic shift in education caused by the pandemic and P 12 school systems made accommodations to support teacher education candidates The pandemic disrupted all established systems and norms however many practices and strategies emerged in educator preparation programs that will have a lasting positive impact on P 20 education and teacher education practices Such practices include the reevaluation of schooling practices with shifts in engagement strategies instructional approaches technology utilization and supporting students and their families Redefining Teacher Education and Teacher Preparation Programs in the Post COVID 19 Era provides relevant innovative practices implemented across teacher education programs and P 20 settings including delivery models training procedures theoretical frameworks district policies and guidelines state national and international standards digital design and delivery of content and the latest empirical research findings on the state of teacher education preparation The book showcases best practices used to shape and redefine teacher education through the COVID 19 pandemic Covering topics such as online teaching practices simulated teaching experiences and emotional learning this text is essential for preservice professionals paraprofessionals administrators P 12 faculty education preparation program designers principals superintendents researchers students and academicians **Inquiry Into Physical Science** Willard J. Jacobson,1969

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