

CHEMISTRY

Gases Full Unit

Full Unit POWERPOINT

Dalton's Law of partial pressure's

The pressure of each gas in a mixture is called the partial pressure of that gas.
Dalton's law of partial pressures states that the total pressure is found by adding the partial pressures of the gases.

$$P_T = P_A + P_B + \dots$$



The Ideal Gas Law

$$PV = nRT$$

New symbols: n and R

Applications of the Ideal Gas Law Molar Mass of an unknown

- Using the ideal gas law to determine the molar mass of a gas
- First Solvent
- The

using $n = PV / RT$
molar mass using:
 $\text{Moles} = m / n$
in the problem

63 TOTAL SLIDES!
2 FULL WEEKS OF
LESSONS!



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Marco Cascella



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