

Chapter 2 Review Sheet

1. Define chemical element, and list the four elements that form the bulk of body matter.
An element is made of just one kind of atom.
Hydrogen, Oxygen Nitrogen, Carbon (HONC)
2. List the subatomic particles, and describe their relative charges and positions in the atom. Proton-positive charge; Neutron-neutral; electron-negative charge. Protons and neutrons comprise the nucleus. Electrons orbit around the nucleus of the atom.
3. Recognize that chemical reactions involve the interaction of electrons to make and break chemical bonds.
4. Describe covalent, ionic and hydrogen bonds.
Covalent bonds are when atoms share electrons in their valence or outer shell. Ionic bonds are when atoms transfer electrons from their valence or outer shell. Hydrogen bonds are when hydrogen atoms bond with a highly electronegative atom such as oxygen. Hydrogen bonds hold water molecules together.
5. Contrast the relative strengths of covalent, ionic and hydrogen bonds. Covalent-Very strong; Ionic- Strong, but not as strong as covalent bonds; hydrogen-very weak bond
6. Describe a chemical reaction. Be able to recognize the reactants and products. A process in which one or more substances, the reactants, are converted to one or more different substances, the products. Substances are either **chemical** elements or compounds.
7. Explain the concept of pH and the pH scale. State the pH of pure water, blood, hydrochloric acid. **pH** is a measure of how acidic/basic water is. The range goes from 0 - 14, with 7 being neutral. pHs of less than 7 indicate acidity, whereas a **pH** of greater than 7 indicates a base. **pH** is really a measure of the relative amount of free hydrogen and hydroxyl ions in the water. Water = 7; blood = 7.4; Hydrochloric acid = 1-3.
8. Describe the function of carbohydrates. Carbohydrates function as an energy source. Give an example of a monosaccharide, disaccharide and a polysaccharide.
Monosaccharide-glucose; Disaccharide-Sucrose; Polysaccharide-starch
9. Draw a monosaccharide, disaccharide and polysaccharide molecule. See ppt or worksheets.
10. Describe the function of lipids. Lipids function as a long term energy source as well as insulation and protection.
11. Identify the three main types of lipids and draw their molecular structure.
1)Triglyceride; 2)Phospholipids; 3)Steroids
12. Describe the functions of proteins. They do most of the work in cells and are required for the structure, **function**, and regulation of the body's tissues and organs. ... Enzymes carry out almost all of the thousands of chemical reactions that take place in cells.

Review Chemistry Of Life Answers

SJ Ball



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