

Hydrocarbons – Physical and Chemical Properties

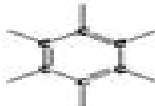
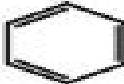
Objective:

In this Lab, we will investigate some of the chemical and physical properties of hydrocarbons and their derivatives and use their properties to distinguish between: alkanes, alkenes, and aromatics

Discussion:

Hydrocarbons are the simplest type of organic compound and they contain only carbon and hydrogen atoms in their structures. Hydrocarbons are classified into different groups based on their chemical bonding pattern. For example: hydrocarbons with only single bonds are known as saturated hydrocarbons (or alkanes) and the ones with multiple bonds (double and triple bonds) are known as unsaturated hydrocarbons (alkenes or alkynes). The detailed classification of saturated and unsaturated hydrocarbons is given in the table below.

Types of hydrocarbons:

Type	Characteristic Functional Group	Simplest Member
Alkane		 Methane (Line-Angle not possible)
Cycloalkane (ring of carbons)		 Cyclopropane
Alkene (has double bond)		 Ethene
Alkyne (has triple bond)		 Ethyne
Aromatic (several double bonds in a ring)		 Benzene

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