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web there are two main rules to remember when balancing nuclear equations 1 the total of the superscripts mass numbers in the reactants and products must be the same 2 the total of the subscripts atomic numbers in the reactants and products must be the same p a r t a what is the value of in the following nuclear reaction

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uranium 238 with a neutron and then spontaneously converts into plutonium 239

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web nuclear reactions are going on all around us using correctly balanced equations is important when trying to understand nuclear reactions all equations need to be balanced to conform to two conservation laws the mass number is conserved and the electrical charge of the nucleus is conserved