

Reversible Reactions and Chemical Equilibrium Worksheet

Supporting questions

- 1) Can you draw the reversible arrow symbol?
- 2) What do you observe when you heat up hydrated copper sulfate?
- 3) Define the term dynamic equilibrium?
- 4) Why can dynamic equilibrium only occur in a closed system?
- 5) What are the 3 factors that alter the position of equilibrium?
- 6) What happens to the position of equilibrium if you add a catalyst?
- 7) For the chemical reaction below can you say what happens to the concentration of ammonia if you (a) Add in more hydrogen (b) increase the pressure (c) increase the temperature (the forward reaction is **exothermic**).



Nitrogen Hydrogen Ammonia

- 8) Why might a high temperature still be used?

Thanks for taking the time to complete these questions, you can check your answers on the next page.

Reversible Reactions And Equilibrium Workbook Answers

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Reversible Reactions And Equilibrium Workbook Answers:

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Chemical Kinetics: Beyond The Textbook Katja Lindenberg, Ralf Metzler, Gleb Oshanin, 2019-08-27 Processes involving randomly moving particles which

react either upon encounter or via distance dependent reaction rates are ubiquitous in nature A few stray examples are recombination of ions or holes and electrons excitation energy migration and quenching trapping of particles by other species coagulation binding of ligands and proteins to specific sites chemotaxis catalytically induced self propulsion polymerization growth of dendrites or aggregates or nuclei of a new phase Several decades ago it was recognized that the kinetic behavior in some systems with reactions and random transport is strongly affected by many factors which were not taken into account in previous studies These are to name but a few fluctuations in the spatial distributions of the reactants and fluctuations of the reactivity some essentially many particle phenomena effects of anomalous diffusion molecular crowding as well as the internal geometry of the reaction bath Within recent years along with a growing interest in chemical processes occurring in biological systems or cellular environments numerous advances have been made and considerable knowledge has been acquired These seminal contributions are however scattered among many journals and no attempt has been made so far to present a unified picture This book presents a general overview of different contemporary facets of chemical kinetics in a variety of different environments It includes 23 seminal works and reviews on different aspects of reaction processes in chemical physical and biophysical systems both theoretical and experimental [UGC NET Environmental Science 3000 + \[MCQ\] Question Answer E-book](#) DIWAKAR EDUCATION HUB ,2022-02-13 UGC NET Environmental Science Unit Wise Practice Question Answer As Per New Updated Syllabus MCQs Highlights 1 Complete Units Cover Include All 10 Units Question Answer 2 300 Practice Question Answer Each Unit 3 Total 3000 Practice Question Answer 4 Try to take all topics MCQ 5 Include Oriented Most Expected Question Answer 6 As Per the New Updated Syllabus Check Sample Pdf Click On Below Link Download Pdf For Free For More Details Call 7310762592 7078549303

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