

AP® CHEMISTRY 2005 EXAMINATIONS CHEMISTRY SECTION

Question 10 (8 points)

As sample of $\text{C}_2\text{H}_2\text{Cl}_2$ is subjected to continuous infrared light of 2.0 micrometers, continuous heat is produced. The product of this sample is subsequently subjected to gas-liquid chromatography.



Use the graph of $\text{C}_2\text{H}_2\text{Cl}_2$ detector response for gas-liquid chromatography to determine the temperature within $\pm 10^\circ\text{C}$ of the maximum. Then, draw a chemical structure for the major product formed in this reaction.

Answer the two:

- On the detector response graph, draw a vertical line that intersects the maximum detector response. Indicate the temperature within $\pm 10^\circ\text{C}$ of the maximum detector response.
- On the detector response graph, draw a vertical line that intersects the maximum detector response. Indicate the temperature within $\pm 10^\circ\text{C}$ of the maximum detector response.

1 point is earned for each correct answer.

$\text{C}_2\text{H}_2\text{Cl}_2$ reacts readily with H_2O to produce $\text{C}_2\text{H}_4\text{Cl}_2$, as represented by the following equation.



One of the $\text{C}_2\text{H}_4\text{Cl}_2$ isomers produced reacts with H_2O to form a product. Write an equation representing the reaction. Indicate all reactants and products, chemical formulas, and physical states.

One of the $\text{C}_2\text{H}_4\text{Cl}_2$ isomers reacts with H_2O to form a product. Write an equation representing the reaction. Indicate all reactants and products, chemical formulas, and physical states.

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Richard H. Langley, John Moore



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