

**Density Worksheet**  
**Physical Science**  
 **$D = m/V$**

| Densities of Common Substances @ 20°C |                              |  |           |                              |
|---------------------------------------|------------------------------|--|-----------|------------------------------|
| Substance                             | Density (g/cm <sup>3</sup> ) |  | Substance | Density (g/cm <sup>3</sup> ) |
| Oxygen                                | 0.00133                      |  | Aluminum  | 2.70                         |
| Hydrogen                              | 0.000084                     |  | Iron      | 7.87                         |
| Ethanol                               | 0.785                        |  | Copper    | 8.96                         |
| Benzene                               | 0.880                        |  | Silver    | 10.5                         |
| Water                                 | 1.000                        |  | Lead      | 11.34                        |
| Magnesium                             | 1.74                         |  | Mercury   | 13.6                         |
| Salt (sodium chloride)                | 2.16                         |  | Gold      | 19.32                        |

1. The ratio of an object's mass to its \_\_\_\_\_ is called the *density* of the object.
2. A kilogram of lead occupies a much smaller volume than a kilogram of water, because \_\_\_\_\_ has a much higher *density*.
3. For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
  - A. mass = 453 g; volume = 225 cm<sup>3</sup>
  - B. mass = 5.0 g; volume = 10.0 cm<sup>3</sup>
  - C. mass = 26.1 g; volume = 2.0 mL
4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
5. A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

# Physical Science If0236 Answers

**J Dewey**



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