

Name _____ Date _____ Per _____

Periodic Trends Lab

Periodic law states that when the elements are arranged in order of increasing atomic number, there is a periodic repetition of their chemical and physical properties. Elements within the same group have the most similar chemical and physical properties, although not identical. Elements in neighboring groups may also have similar chemical and physical properties, but to a lesser degree. Trends in properties occur within groups and across periods.

PURPOSE: In this activity, you will

- Interpret group and periodic trends in atomic radius, electronegativity, and ionization energy
- Graph these trends

PROCEDURE:

1. Complete the pre-lab section
2. Using the data from **Table S** of your reference tables, prepare the graphs listed below. Staple your graphs to the back of this lab packet.

On each graph be sure to:

- **USE PENCIL!**
- Put an appropriate title and labels for each axis.
- Omit the transition elements.
- Connect the points you plot.
- Connect the points representing Group 1 elements with a green line.
- Connect the points representing Group 2 elements with a blue line.
- Connect the points representing Group 17 elements with a red line.
- Draw a triangle around the points representing the elements of Period 3.

GRAPH 1. Atomic radius vs. atomic number for elements 1-54.

GRAPH 2. Electronegativity vs. atomic number for elements 1-54.

GRAPH 3. Ionization energy vs. atomic number for elements 1-54.

3. Complete the conclusion section.

Periodic Trends Pre Lab Answers

Mike Jess

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

Periodic Trends Pre Lab Answers:

This book delves into Periodic Trends Pre Lab Answers. Periodic Trends Pre Lab Answers is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Periodic Trends Pre Lab Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Periodic Trends Pre Lab Answers
 - Chapter 2: Essential Elements of Periodic Trends Pre Lab Answers
 - Chapter 3: Periodic Trends Pre Lab Answers in Everyday Life
 - Chapter 4: Periodic Trends Pre Lab Answers in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, this book will provide an overview of Periodic Trends Pre Lab Answers. The first chapter will explore what Periodic Trends Pre Lab Answers is, why Periodic Trends Pre Lab Answers is vital, and how to effectively learn about Periodic Trends Pre Lab Answers.
 3. In chapter 2, the author will delve into the foundational concepts of Periodic Trends Pre Lab Answers. This chapter will elucidate the essential principles that must be understood to grasp Periodic Trends Pre Lab Answers in its entirety.
 4. In chapter 3, the author will examine the practical applications of Periodic Trends Pre Lab Answers in daily life. The third chapter will showcase real-world examples of how Periodic Trends Pre Lab Answers can be effectively utilized in everyday scenarios.
 5. In chapter 4, the author will scrutinize the relevance of Periodic Trends Pre Lab Answers in specific contexts. The fourth chapter will explore how Periodic Trends Pre Lab Answers is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Periodic Trends Pre Lab Answers. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Periodic Trends Pre Lab Answers.

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