Unit 2: The Periodic Table Review Sheet

UZLI: PT History

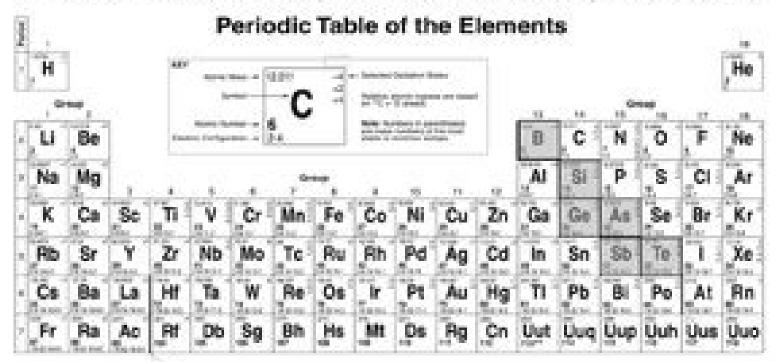
- On the present day Periodic Table, elements are arranged in order of increasing atomic number
- Groups: vertical columns on the periodic table
 - Elements in the same group have similar chemical properties because they have the same number of valence electrons
 - Valence electron: outermost electrons (lost # in the electron configuration)
- Periods: horizontal rows on the periodic table.

U21.2: Parts of the Periodic Table

Elements are classified as either metals, nanmetals, or metalloids.

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Classification can be determined by location on the periodic table or by different properties.



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Periodic Table Review Pt 2

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