

## Percent, Actual, and Theoretical Yield

Example 1:  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$   
 If 10.0 g of  $\text{H}_2$  reacts with 10.0 g of  $\text{O}_2$ , what is the theoretical yield of  $\text{H}_2\text{O}$ ?

$$\begin{aligned} \text{Molar mass of } \text{H}_2 &= 2.02 \text{ g/mol} \\ \text{Molar mass of } \text{O}_2 &= 32.00 \text{ g/mol} \\ \text{Molar mass of } \text{H}_2\text{O} &= 18.02 \text{ g/mol} \end{aligned}$$

$$\begin{aligned} \text{Moles of } \text{H}_2 &= \frac{10.0 \text{ g}}{2.02 \text{ g/mol}} = 4.95 \text{ mol} \\ \text{Moles of } \text{O}_2 &= \frac{10.0 \text{ g}}{32.00 \text{ g/mol}} = 0.3125 \text{ mol} \end{aligned}$$

$$\text{The limiting reagent is } \text{O}_2 \text{ because it has the smallest number of moles.}$$

$$\text{Theoretical yield of } \text{H}_2\text{O} = 0.3125 \text{ mol} \times 18.02 \text{ g/mol} = 5.63 \text{ g}$$

Example 2:  $\text{Fe} + \text{S} \rightarrow \text{FeS}$   
 If 10.0 g of  $\text{Fe}$  reacts with 10.0 g of  $\text{S}$ , what is the theoretical yield of  $\text{FeS}$ ?

$$\begin{aligned} \text{Molar mass of } \text{Fe} &= 55.85 \text{ g/mol} \\ \text{Molar mass of } \text{S} &= 32.07 \text{ g/mol} \\ \text{Molar mass of } \text{FeS} &= 87.92 \text{ g/mol} \end{aligned}$$

$$\begin{aligned} \text{Moles of } \text{Fe} &= \frac{10.0 \text{ g}}{55.85 \text{ g/mol}} = 0.179 \text{ mol} \\ \text{Moles of } \text{S} &= \frac{10.0 \text{ g}}{32.07 \text{ g/mol}} = 0.312 \text{ mol} \end{aligned}$$

$$\text{The limiting reagent is } \text{Fe} \text{ because it has the smallest number of moles.}$$



Example 3:  $\text{Fe} + \text{S} \rightarrow \text{FeS}$   
 If 10.0 g of  $\text{Fe}$  reacts with 10.0 g of  $\text{S}$ , what is the theoretical yield of  $\text{FeS}$ ?

$$\frac{10.0 \text{ g Fe}}{55.85 \text{ g/mol}} = 0.179 \text{ mol Fe}$$

Example 4:  $\text{Fe} + \text{S} \rightarrow \text{FeS}$   
 If 10.0 g of  $\text{Fe}$  reacts with 10.0 g of  $\text{S}$ , what is the theoretical yield of  $\text{FeS}$ ?

$$\begin{aligned} \text{Molar mass of } \text{Fe} &= 55.85 \text{ g/mol} \\ \text{Molar mass of } \text{S} &= 32.07 \text{ g/mol} \\ \text{Molar mass of } \text{FeS} &= 87.92 \text{ g/mol} \end{aligned}$$

$$\begin{aligned} \text{Moles of } \text{Fe} &= \frac{10.0 \text{ g}}{55.85 \text{ g/mol}} = 0.179 \text{ mol} \\ \text{Moles of } \text{S} &= \frac{10.0 \text{ g}}{32.07 \text{ g/mol}} = 0.312 \text{ mol} \end{aligned}$$

$$\text{The limiting reagent is } \text{Fe} \text{ because it has the smallest number of moles.}$$

$$\text{Theoretical yield of } \text{FeS} = 0.179 \text{ mol} \times 87.92 \text{ g/mol} = 15.7 \text{ g}$$

# Percent Actual And Theoretical Yield Answer Key

**Xiaolong Qi**



## **Percent Actual And Theoretical Yield Answer Key:**

**Introduction to General, Organic, and Biochemistry** Morris Hein, Scott Pattison, Susan Arena, Leo R. Best, 2014-01-15

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**Introduction to Chemical Principles** Howard Stephen Stoker, 1986 Chemical Principles, Properties, and Reactions

Kenneth L. Henold, Frank Walmsley, 1984 *5 Steps to a 5: AP Chemistry 2018 Elite Student Edition* Mary Millhollon, Richard H. Langley, 2017-07-28 Get ready to ace your AP Chemistry Exam with this easy to follow multi platform study guide 5 Steps to a 5 AP Chemistry 2018 Elite Student Edition introduces an effective 5 step study plan to help you build the skills knowledge and test taking confidence you need to achieve a high score on the exam This popular test prep guide matches the latest course syllabus and latest exam You ll get online help four full length practice tests two in the book and two online detailed answers to each question study tips and important information on how the exam is scored Because this guide is accessible in print and digital formats you can study online via your mobile device straight from the book or any combination of the three With the new 5 Minutes to a 5 section you ll also get an extra AP curriculum activity for each school day to help reinforce the most important AP concepts With only 5 minutes a day you can dramatically increase your score on exam day 5 Steps to a 5 AP Chemistry 2018 Elite Student Edition features New 5 Minutes to a 5 Concise activities reinforcing the most important AP concepts and presented in a day to day study format Access to the entire Cross Platform Prep Course in Chemistry 4 Practice Exams 2 in the book 2 online Powerful analytics you can use to assess your test readiness Flashcards games social media support and more **5 Steps to a 5 AP Chemistry, 2014-2015 Edition** Richard H. Langley, John Moore, 2013-07-10 A PERFECT PLAN for the PERFECT SCORE STEP 1 Set up your study plan with three customized study schedules STEP 2 Determine your readiness with an AP style diagnostic exam STEP 3 Develop the strategies that will give you the edge on test day STEP 4 Review the terms and concepts you need to score high STEP 5 Build your confidence with full length practice exams 5 Steps to a 5: AP Chemistry 2018 Mary Millhollon, Richard H. Langley, 2017-07-28 Get ready to ace your AP Chemistry Exam with this easy to follow multi platform study guide 5 Steps to a 5 AP Chemistry introduces an easy to follow effective 5 step study plan to help you build the skills knowledge and test taking confidence you need to achieve a high score on the exam This wildly popular test prep guide matches the latest course syllabus and the latest exam You ll get online help four full length practice tests two in the book and two online detailed answers to each question study tips information on how the exam is scores and much more Because this guide is accessible in print and digital formats you can study online via your mobile device straight from the book or any combination of the three 5 Steps to a 5 AP Chemistry 2018 features New Access to the entire Cross Platform Prep Course in Chemistry 4 Practice Exams 2 in the book 2 online An interactive customizable AP Planner app to help you organize your time Powerful analytics you can use to assess your test readiness Flashcards games and more **5 Steps to a 5: AP Chemistry 2017** Mary Millhollon, Richard H. Langley, 2016-07-29 Get ready for your AP Chemistry exam with this straightforward easy to follow study guide updated to match the latest test changes The wildly popular test prep guide updated and enhanced for smartphone users 5 Steps to a 5 AP Chemistry 2017 provides a proven strategy to achieving high scores on this demanding Advanced Placement exam This logical and easy to follow instructional guide introduces an effective 5 step study plan to help students build the skills

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