

| Page 3 | Mark Scheme: Teachers' version IGCSE – May/June 2011 | Syllabus 0620 | Paper 31 |
|--------|---|------------------|--|
| | (c) base not alkali accepts a proton accepts hydrogen ion / H^+ only [1] proton and H^+ [2] | | [1] [2] |
| 3 | (a) any four max 4 carbon forms carbon dioxide / carbon monoxide this is a gas it escapes / blown out / diffuses silicon forms silicon(IV) oxide / silica / silicon(IV) oxide present in impure iron silicon(IV) oxide reacts with calcium oxide to form slag or calcium silicate slag removed from surface accept skimmed, syphoned, poured off not tapped accept correct formula or equations not calcium oxide reacts with silicon | | [1] [1] [1] [1] [1] max [4] |
| | (b) (i) any sensible suggestion – harder/stronger/can be tailored for a specific use/more resistant to corrosion not steel does not rust | | [1] |
| | (ii) mild steel – cars or any vehicle/bicycles/white goods/screws or nails/poll/bridges/boots/buildings/ships/pipes/machinery etc. stainless steel – chemical plants/cooking utensils/jewellery/cutlery/surgical equipment/kitchen sinks/pipes/etc. | | [1] [1] |
| | (c) (i) strong attractive forces / strong bonds / bonds hard to break / requires a lot of energy to break bonds not between ions, not between positive and negative ions, not between electrons between positive ions and (negative) electrons / opposite charges attract | | [1] [1] |
| | (ii) because the layers, lattice or rows of ions/cations accept sheets of ions not atoms / molecules / protons / nuclei can move / slip / slide past each other | | [1] [1] |
| 4 | (a) (i) $2ZnS + 3O_2 \rightarrow 2ZnO + 2SO_2$ not balanced only [1] | | [2] |
| | (ii) two reagents from named metal(s) more reactive than zinc/carbon monoxide not hydrogen | | [2] |
| | (iii) they have different boiling points cadmium will distil first then zinc leaving lead/lead distilled last | | [1] [1] |

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Locatelli, Sebastian Granica**



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1. Understanding the eBook November 2011 0620 31 Ms
 - The Rise of Digital Reading November 2011 0620 31 Ms
 - Advantages of eBooks Over Traditional Books
2. Identifying November 2011 0620 31 Ms
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an November 2011 0620 31 Ms
 - User-Friendly Interface
4. Exploring eBook Recommendations from November 2011 0620 31 Ms

- Personalized Recommendations
 - November 2011 0620 31 Ms User Reviews and Ratings
 - November 2011 0620 31 Ms and Bestseller Lists
5. Accessing November 2011 0620 31 Ms Free and Paid eBooks
- November 2011 0620 31 Ms Public Domain eBooks
 - November 2011 0620 31 Ms eBook Subscription Services
 - November 2011 0620 31 Ms Budget-Friendly Options
6. Navigating November 2011 0620 31 Ms eBook Formats
- ePub, PDF, MOBI, and More
 - November 2011 0620 31 Ms Compatibility with Devices
 - November 2011 0620 31 Ms Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of November 2011 0620 31 Ms
 - Highlighting and Note-Taking November 2011 0620 31 Ms
 - Interactive Elements November 2011 0620 31 Ms
8. Staying Engaged with November 2011 0620 31 Ms
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers November 2011 0620 31 Ms
9. Balancing eBooks and Physical Books November 2011 0620 31 Ms
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection November 2011 0620 31 Ms
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine November 2011 0620 31 Ms
- Setting Reading Goals November 2011 0620 31 Ms
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of November 2011 0620 31 Ms

- Fact-Checking eBook Content of November 2011 0620 31 Ms
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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