

Chemical Formula

Talking about proportions, molecular and relative of atoms indicated by a subscript.

Molecular: $C_{10}H_{12}O_4$ (molecular)

Formula represents the number of _____ of _____
in a single molecule.

Empirical: $C_5H_6O_2$ (molecular)

Formula represents the proportions of _____ of
positive and negative _____, hence formula
unit.

Atomic Mass

The proportions of the atoms relative to hydrogen
unit together.

Relative

$\frac{1}{12}$ mass of carbon atom relative to hydrogen

Relative mass proportionally charged
unit

Subscript

$\frac{1}{12}$ mass of carbon atom

Relative mass proportionally charged

$\frac{1}{12}$ mass

Note Taking Guide Episode 6chemical Formula

G Thomas



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Lovejoy's College Guide, 1993 **Resources in Education**, 1997-04 **Railway Locomotives and Cars**, 1860
The Athenaeum, 1906 **The Nation**, 1893 Southern Dental Journal and Luminary, 1883 **Athenaeum and Literary Chronicle**, 1854 **Medical Record** George F. Shrady, Thomas Lathrop Stedman, 1921 The Saturday Evening Post, 1909 **Scientific American**, 1897 **Nature** Sir Norman Lockyer, 1912 **Forthcoming Books** Rose
Army, 1996-06 *British Books*, 1910 **ENC Update** Eisenhower National Clearinghouse for Mathematics and Science Education, 1995 This publication is designed to tap into fresh stories and ideas about mathematics and science teachers who are charting new territory in education **Journal of Education**, 1884 *Billboard*, 1953-07-04 In its 114th year Billboard remains the world's premier weekly music publication and a diverse digital events brand content and data licensing platform Billboard publishes the most trusted charts and offers unrivaled reporting about the latest music video gaming media digital and mobile entertainment issues and trends The Academy, 1884 Outlook Alfred Emanuel Smith, Francis Walton, 1878
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