### temperature:

- a measure of the <u>average kinetic</u> energy of the particles in a sample of matter
- does not depend on the amount of matter in the sample
- symbol is <u>⊤</u> : unit is <u>c</u>

### heat:

- total amount of kinetic energy that flows because of a difference in tempature.
- depends on mass \_\_\_\_\_ of sample

Kinetic energy is energy in motion

Potential energy is store

Potential energy is hiding and cannot be

measured

Only <u>changes</u> in P.E. can be measured.

### specific heat capacity:



When heat (Q) is absorbed by a system, part of it (C) goes into storage as \_\_\_\_\_\_ energy and part of it is used to make the molecules move around \_\_\_\_\_ , raising the \_\_\_\_\_ (\Delta t).

Why does sand get hotter in the day and colder at night than the water?

# **Note Taking Guide Episode 1302 Chemistry Key**

**L** Manion

Note Taking Guide Episode 1302 Chemistry Key:

## Note Taking Guide Episode 1302 Chemistry Key Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has be apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "Note Taking Guide Episode 1302 Chemistry Key," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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