

Periodic Table of the Elements

	+1	+2											+3	+4	+5	+6	+7	0						
1	H 1.008	He 4.003											B 10.81	C 12.01	N 14.01	O 16.00	F 18.99	Ne 20.18						
2	Li 6.941	Be 9.012											Al 26.98	Si 28.09	P 30.97	S 32.07	Cl 35.45	Ar 39.95						
3	Na 22.99	Mg 24.31	Sc 44.96	Ti 47.88	V 50.94	Cr 52.00	Mn 54.94	Fe 55.85	Co 58.93	Ni 58.71	Cu 63.55	Zn 65.38	Ga 69.72	Ge 72.64	As 74.92	Se 78.96	Br 79.90	Kr 83.80						
4	K 39.10	Ca 40.08	Sc 44.96	Ti 47.88	V 50.94	Cr 52.00	Mn 54.94	Fe 55.85	Co 58.93	Ni 58.71	Cu 63.55	Zn 65.38	Ga 69.72	Ge 72.64	As 74.92	Se 78.96	Br 79.90	Kr 83.80						
5	Rb 85.47	Sr 87.62	Y 88.91	Zr 91.22	Nb 92.91	Mo 95.94	Tc 98.91	Ru 101.07	Rh 102.91	Pd 106.42	Ag 107.87	Cd 112.41	In 114.82	Sn 118.71	Sb 121.76	Te 127.60	I 126.91	Xe 131.29						
6	Cs 132.91	Ba 137.33	La 138.91	Ce 140.12	Pr 140.91	Nd 144.24	Pm 144.91	Sm 150.36	Eu 151.96	Gd 157.25	Tb 158.93	Dy 162.50	Ho 164.93	Er 167.26	Tm 168.93	Yb 173.05	Lu 174.97							
7	Fr [223]	Ra [226]	Ac [227]	Th [232]	Pa [231]	U [238]	Np [237]	Pu [244]	Am [243]	Cm [247]	Bk [247]	Cf [251]	Es [252]	Fm [257]	Mn [288]	Co [285]	Ni [285]	Pu [289]	Am [287]	Cm [284]	Bk [289]	Cf [287]	Es [285]	Fm [283]
8	Ce [140.12]	Pr [140.91]	Nd [144.24]	Pm [144.91]	Sm [150.36]	Eu [151.96]	Gd [157.25]	Tb [158.93]	Dy [162.50]	Ho [164.93]	Er [167.26]	Tm [168.93]	Yb [173.05]	Lu [174.97]										
9	Th [232]	Pa [231]	Np [237]	Pu [244]	Am [243]	Cm [247]	Bk [247]	Cf [251]	Es [252]	Fm [257]	Mn [288]	Co [285]	Ni [285]	Pu [289]	Am [287]	Cm [284]	Bk [289]	Cf [287]	Es [285]	Fm [283]				

The only metal that is liquid at room temperature.

Period 3
element with
7 valence
electrons.

Semi-conductor
metalloid used
in electronics.

Only one valence electron, but not part of a family.

Extremely reactive non-metal.

Group that will form 2+ cations.

The only non-metal that is liquid at room temperature.

Completely
unreactive family
with complete
valence shell.

Extremely reactive metal.

Group with 4 valence electrons.

Periodic Trends Review

CH Cherryholmes

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