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# The Periodic Law

## Section Quiz: History of the Periodic Table

As Marjorie completed each question of this quiz, she placed that question's number statement in her answer's box to question.

1. In developing his periodic table, Mendeleev's observations were
  - a. periodic elements.
  - b. elements with liquid state.
  - c. isotopes.
  - d. **properties.**
2. Mendeleev's periodic table showed that all elements in each of the columns shared many features, but grouped together elements with
  - a. **properties.**
  - b. similar elements.
  - c. isotopes.
  - d. isotopes.
3. Mendeleev predicted that the gaps in his periodic table represented
  - a. ions.
  - b. unknown elements.
  - c. unstable elements.
  - d. **undiscovered elements.**
4. The periodic table came before a periodic table based on the
  - a. number of protons.
  - b. **atomic weight.**
  - c. isotopes.
  - d. mass number.
5. The periodic law describes the physical and chemical properties of
  - a. **elements.**
  - b. isotopes.
  - c. ions.
  - d. elements.

# Periodic Law Section Quiz Answers

**Adrian Dingle**



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