

Types of Chemical Reactions Lab Answer Key

Chemical Reaction #1

| | |
|--|--|
| <i>Reactants (write in words)</i> | hydrochloric acid and magnesium |
| <i>Observations</i> | accept most responses |
| <i>Type of Reaction</i> | single replacement |
| <i>Predict Products (write in words)</i> | magnesium chloride and hydrogen gas |
| <i>Equation (write in symbols)</i> | $2\text{HCl}_{\text{(aq)}} + \text{Mg}_{\text{(s)}} \rightarrow \text{MgCl}_{2\text{(aq)}} + \text{H}_{2\text{(g)}}$ |

Chemical Reaction #2

| | |
|--|---|
| <i>Reactants (write in words)</i> | copper metal and oxygen gas |
| <i>Observations</i> | accept most responses |
| <i>Type of Reaction</i> | synthesis |
| <i>Predict Products (write in words)</i> | copper oxide |
| <i>Equation (write in symbols)</i> | $2\text{Cu}_{\text{(s)}} + \text{O}_{2\text{(g)}} \rightarrow 2\text{CuO}_{\text{(s)}}$ |

Chemical Reaction #3

| | |
|--|--|
| <i>Reactants (write in words)</i> | methane gas and oxygen gas |
| <i>Observations</i> | accept most responses |
| <i>Type of Reaction</i> | combustion |
| <i>Predict Products (write in words)</i> | carbon dioxide and water |
| <i>Equation (write in symbols)</i> | $\text{CH}_{4\text{(g)}} + 2\text{O}_{2\text{(g)}} \rightarrow \text{CO}_{2\text{(g)}} + 2\text{H}_2\text{O}_{\text{(g)}}$ |

Chemical Reaction #4

| | |
|--|---|
| <i>Reactants (write in words)</i> | hydrogen peroxide |
| <i>Observations</i> | accept most responses |
| <i>Type of Reaction</i> | decomposition |
| <i>Predict Products (write in words)</i> | water and oxygen gas |
| <i>Equation (write in symbols)</i> | $2\text{H}_2\text{O}_{2\text{(l)}} \rightarrow 2\text{H}_2\text{O}_{\text{(l)}} + \text{O}_{2\text{(g)}}$ |

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