#### Atomic Number

- \_\_\_\_\_ across a period
- \_\_\_\_\_ down a family

### Atomic Mass

- \_\_\_\_\_ across a period
- \_\_\_\_\_ down a family

### Atomic Radius

\_\_\_\_ the distance between \_\_\_\_\_ of two like \_\_\_\_\_.

- \_\_\_\_\_ across a period (due to increased attraction of more positive nuclei to electron clouds)
- \_\_\_\_\_\_ down a family (due to addition of an energy level)

## Electron Affinity

amount of \_\_\_\_\_ gains an

- \_\_\_\_\_ across a period
- \_\_\_\_\_ down a family

### Ionization Energy

amount of \_\_\_\_\_ required to \_\_\_\_\_ an \_\_\_\_ from a neutral \_\_\_\_

- across a period
- \_\_\_\_\_ down a family

### Electron Distribution - Forming Ions

- Atoms become ions by \_\_\_\_\_ or \_\_\_\_\_ electrons to become.
- An atom is stable when it has a \_\_\_\_\_ outer energy level (\_\_\_\_ valence electrons).
- Atoms that lose (give away) electrons become \_\_\_\_\_ charged ions.
   Atoms that gain (steal) electrons become \_\_\_\_\_ charged ions.

### The Chemistry Quiz

CR1. CR2. 1. 2. 3. 4. 5.

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**Brian Clegg** 

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