

Background

- 1. Study of this \_\_\_\_\_ multi \_\_\_\_\_ relationship is a \_\_\_\_\_ method \_\_\_\_\_
- 2. Research on \_\_\_\_\_ relationship \_\_\_\_\_ equated \_\_\_\_\_

2.  $40g = 40r + 40g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide

The \_\_\_\_\_ is a \_\_\_\_\_ reaction \_\_\_\_\_ give the \_\_\_\_\_ \_\_\_\_\_ from the \_\_\_\_\_ involved in the reaction \_\_\_\_\_

3. Problem

When aluminum chloride reacts with aluminum hydroxide, aluminum chloride is produced



Write the balanced chemical equation for the reaction



2. If you start with 4 moles of  $Al$ , how many moles of  $Al(OH)_3$  will be produced?

2 moles  $Al(OH)_3$  and 3 moles  $H_2O$  and 3 moles  $HCl$

# Note Taking Guide Episode 801

**Lingjun Ying**



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ml 4 6 g 100 ml

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