

1. Calculate the energy of the yellow light that has wavelength of 589nm(Planck constant,
 -34
 $h=6.626 \times 10^{-34} \text{ J-s}$

2. What is the wave length of an electron that moving with speed of $5.97 \times 10^6 \text{ m/s}$ and the mass
 -31 -34
of electron is $9.11 \times 10^{-31} \text{ kg}$ (Planck constant, $h= 6.626 \times 10^{-34} \text{ J-s}$.

3. Define and example of transitional metal, write the condensed electro configuration of As.

4. Define and example of metal Oxide, mention 2 properties of metal Oxide.

5. Write down 4 rules to write the Lewis structure, what is bond polarity?

6. Write the number of valence electron of PCl_3 and BrO_3^- ion .

Scc201 Tutorial 3

Lauren Pecorino



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