

Max. Marks:70

Time: 3 hours

General Instructions:

Read the following instructions carefully.

- (a) There are 33 questions in this question paper with internal choice.
 - (b) SECTION A consists of 16 multiple -choice questions carrying 1 mark each.
 - (c) SECTION B consists of 5 short answer questions carrying 2 marks each.
 - (d) SECTION C consists of 7 short answer questions carrying 3 marks each.
 - (e) SECTION D consists of 2 case - based questions carrying 4 marks each.
 - (f) SECTION E consists of 3 long answer questions carrying 5 marks each.
 - (g) All questions are compulsory.
 - (h) Use of log tables and calculators is not allowed.
-

SECTION A

The following questions are multiple -choice questions with one correct answer. Each question carries 1 mark. There is no internal choice in this section.

1. Which of the following solutions will have the highest conductivity at 298 K?

(a) 0.01 M HCl solution	(b) 0.1 M HCl solution
(c) 0.01 M CH ₃ COOH solution	(d) 0.1 M CH ₃ COOH solution



Identify A and B:

- | | |
|--|---|
| (a) A = 1-phenylethanal , B = acetophenone | (b) A = Benzophenone B = formaldehyde |
| (c) A = Benzaldehyde , B = Acetophenone | (d) A = Benzophenone , B = Acetophenone |
3. The vitamins which can be stored in our body are:

(a) Vitamin A, B, D and E	(d) Vitamin A, C, D and K
(c) Vitamin A, B, C and D	(d) Vitamin A, D, E and K
 4. What is IUPAC name of the ketone A, which undergoes iodo form reaction to give CH₃CH= C(CH₃)COONa and yellow precipitate of CHI₃?

(a) 3-Methylpent-3-en-2-one	(b) 3-Methylbut-2-en- one
(c) 2, 3-Dimethylethanone	(d) 3-Methylpent-4-one
 5. Which of the following is not correct?

(a) In haloarenes, the electron pairs on halogen atom are in conjugation with π -electrons of the ring.
(b) The carbon-magnesium bond is covalent and non-polar in nature .
(c) During S _N ¹ reaction, the carbocation formed in the slow step being sp ² hybridised is planar.
(d) Out of CH ₂ = CH-Cl and C ₆ H ₅ CH ₂ Cl, C ₆ H ₅ CH ₂ Cl is more reactive towards S _N ¹ reaction

Sample Question Paper Applied Chemistry

Katrin Zwirglmaier



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