

Periods and Groups: Valence Electrons: Memory

Periods and Groups

Elements that have the same number of valence electrons have similar properties, so we put them in groups. The the periodic table groups + columns that look like this. They will see family groups 1A-18A. Hydrogen is in the same period as Helium.

Elements that have electrons in the outer shell have similar properties in the Periodic Table. Periods go left to right. Elements in the same vertical columns

Period	Group		Group					Group	
	1A	2A	3A	4A	5A	6A	7A	8A	1A
1	H	He							He
2	Li	Be							Ne
3	Na	Mg							Ar
4	K	Ca							Kr

Thinking like the game "Memory", answer that an element in the periodic group _____

What element is in Group 1, and Group 18A? Group 2 element is Group 18A is Group 18A is _____

What element is in Group 1A and Period 2?

What element is in Group 2A and Period 3?

What group number is in Group 1A?

What group number is in Group 18A?

Valence Electrons

Valence Electrons are the most important electrons in an atom. Each group contains has the same number of valence electrons. Valence electrons are the electrons that are involved in chemical bonding.

Hydrogen 1



1 valence electron

Helium 2



2 valence electrons

Lithium 3



3 valence electrons

Hydrogen and Helium are both in Group 1A. Helium is 2 valence electrons. Hydrogen is 1 valence electron.

Group 1A, Helium 2A and Group 18A have 2 valence electrons. How many valence electrons does Group 1A have?

How many valence electrons?

Group 1A - Elements have 1 valence electron. Group 2A - Elements have 2 valence electrons. Group 3A - Elements have 3 valence electrons. Group 4A - Elements have 4 valence electrons. Group 5A - Elements have 5 valence electrons. Group 6A - Elements have 6 valence electrons. Group 7A - Elements have 7 valence electrons. Group 8A - Elements have 8 valence electrons.

Periods And Groups Valence Electrons Masses Answers

Dr. Amar Bhutani



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