

PERIODIC TABLE

Atomic Properties of the Elements

NIST
National Institute of
Standards and Technology
U.S. Department of Commerce

Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data					2					
For the most accurate values of these and other constants, visit physics.nist.gov/constants																		physics.nist.gov		www.nist.gov/srd					He					
1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³ Cs																		13	14	15	16	17	18							
																		IIIA	IVA	VA	VIA	VIIA	VIIIA							
1	H																	5	B	6	C	7	N	8	O	9	F	10	Ne	
2	Li	Be																	11	Al	12	Si	13	P	14	S	15	Cl	16	Ar
3	Na	Mg	3	4	5	6	7	8	9	10	11	12																		
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr												
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe												
6	Cs	Ba			Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn											
7	Fr	Ra			Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Uuq	Uup	Uuh	Uus	Uuo											

Atomic Number: 58
Ground-state Ionization Energy (eV): 5.5337

Symbol: **Ce**
Name: Cerium
Atomic Weight: 140.116

Ground-state Configuration: [Xe]4f¹5d¹6s²
Ionization Energy (eV): 5.5337

*Based upon ¹²C. () indicates the mass number of the longest-lived isotope.

For a description of the data, visit physics.nist.gov/data

NIST SP 966 (September 2010)

Periodix Table Study Packet

Brendan G. Carr

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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