

Chemistry I: Periodicity Worksheet

- 1) Arrange the following elements in order of increasing size: Al, Br, C, K, and Na

100

- 2) Arrange the following elements in order of increasing size: Ca, Rb, P, Ge, and Sr.

100

33. Circle the atom or ion in each pair that has the larger radius.

- ✓ **Immer!**

- 4) Select the atom or ion in each pair that has the larger radius.

- ch. Error As

- 5) Which of the following groups of elements is arranged correctly in order of increasing ionization energy?
(Answer by circling the correct answer.)

- d) $\text{Ne} < \text{C} < \text{Si} < \text{Li}$

- 6) Arrange the following atoms in order of increasing ionization energy: Li, K, C, and N.

100

- 7) Arrange the following atoms in order of increasing ionization energy: Si, K, As, and Ca.

100

- 8) Compare the elements Na, Mg, O, and P.

-

Periodicity Worksheet Answer Key

Richard Bailey

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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