

Periodic Trends – Atomic Mass

- The **atomic mass** describes the mass of each individual element.
- **Atomic mass** increases moving **down** each group and **right** across a period.

Instructions: In the periodic table below, draw two arrows representing the direction that atomic mass **increases** across groups and periods.

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3			3	4	5	6	7	8	9	10	11	12						
4																		
5																		
6																		
7																		

Instructions: In each question, identify the element with the **greatest** atomic mass:

- 1) Chlorine (Cl), Iodine (I), Bromine (Br), Fluorine (F) Iodine
- 2) Oxygen (O), Carbon (C), Nitrogen (N), Boron (B) Oxygen
- 3) Lithium (Li), Silicon (Si), Sulfur (Br), Xenon (Xe) Xenon

Instructions: In each question, list the elements from **least** to **greatest** atomic mass:

- 4) Magnesium (Mg), Beryllium (Be), Barium (Ba) Be, Mg, Ba
- 5) Gallium (Ga), Selenium (Se), Potassium (K) K, Ga, Se

Periodic Trends In Atomic Properties Answers

J Ma



Periodic Trends In Atomic Properties Answers:

This book delves into Periodic Trends In Atomic Properties Answers. Periodic Trends In Atomic Properties Answers is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Periodic Trends In Atomic Properties Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Periodic Trends In Atomic Properties Answers
- Chapter 2: Essential Elements of Periodic Trends In Atomic Properties Answers
- Chapter 3: Periodic Trends In Atomic Properties Answers in Everyday Life
- Chapter 4: Periodic Trends In Atomic Properties Answers in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Periodic Trends In Atomic Properties Answers. This chapter will explore what Periodic Trends In Atomic Properties Answers is, why Periodic Trends In Atomic Properties Answers is vital, and how to effectively learn about Periodic Trends In Atomic Properties Answers.
3. In chapter 2, this book will delve into the foundational concepts of Periodic Trends In Atomic Properties Answers. The second chapter will elucidate the essential principles that must be understood to grasp Periodic Trends In Atomic Properties Answers in its entirety.
4. In chapter 3, the author will examine the practical applications of Periodic Trends In Atomic Properties Answers in daily life. This chapter will showcase real-world examples of how Periodic Trends In Atomic Properties Answers can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Periodic Trends In Atomic Properties Answers in specific contexts. This chapter will explore how Periodic Trends In Atomic Properties Answers is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Periodic Trends In Atomic Properties Answers. This chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Periodic Trends In Atomic Properties Answers.

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