

Name: _____ KEY _____ Period _____ Date: _____

Fill in the blank (2 points each blank)

1. Elements that have the properties of both metal and non-metals are metalloids.
2. What do elements in the same group have the same number of? Valence electrons
3. Which element would have the highest electron affinity? Fluorine
4. What element is at period 3, group 8A? Argon
5. How many valence electrons does Bromine have? 7
6. The atomic number tells you how many protons (and electrons if neutral) there are in an atom.
7. Two Characteristics of Non-metals are not good conductors of heat or electricity, not malleable, etc.
8. Two characteristics of Metals are malleable, ductile, good conductors of heat and electricity
9. Which element would have the smallest atomic radius? Helium
10. Which element would have the highest ionization energy? Helium
11. For the questions below and based on the information provided, fill in the missing information. (1 point each)

Name	Symbol	# of protons	# of electrons	# of neutrons	Atomic Number	Mass Number
Potassium	K	19	19	22	19	41
Phosphorous	P	15	15	18	15	33
Beryllium	Be	4	4	4	4	8
Sulfur	S	16	16	20	16	36

12. Draw Bohr Models for (3 points each)

a. Na23

b. N15

13. What is an isotope? (1 point)

An atom with a different number of neutrons

14. How can you tell the identity of any atom? (1 point)

Number of PROTONS

15. Alien Periodic Chart -- Directions: Read the clue and place the alien symbol in the corresponding box!

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The Princeton Review



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Table of Contents Periodic Table Test A Answer Key

1. Understanding the eBook Periodic Table Test A Answer Key
 - The Rise of Digital Reading Periodic Table Test A Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Test A Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Test A Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Test A Answer Key
 - Personalized Recommendations
 - Periodic Table Test A Answer Key User Reviews and Ratings
 - Periodic Table Test A Answer Key and Bestseller Lists
5. Accessing Periodic Table Test A Answer Key Free and Paid eBooks
 - Periodic Table Test A Answer Key Public Domain eBooks
 - Periodic Table Test A Answer Key eBook Subscription Services

- Periodic Table Test A Answer Key Budget-Friendly Options
- 6. Navigating Periodic Table Test A Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Test A Answer Key Compatibility with Devices
 - Periodic Table Test A Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Test A Answer Key
 - Highlighting and Note-Taking Periodic Table Test A Answer Key
 - Interactive Elements Periodic Table Test A Answer Key
- 8. Staying Engaged with Periodic Table Test A Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Test A Answer Key
- 9. Balancing eBooks and Physical Books Periodic Table Test A Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Test A Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Test A Answer Key
 - Setting Reading Goals Periodic Table Test A Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Test A Answer Key
 - Fact-Checking eBook Content of Periodic Table Test A Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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