# 2.6 - Periodic Trends worksheet

Name	
I. Short answer	
1. Label each group on the periodic tab	de.
<ol><li>Explain what valence electrons are.</li></ol>	
3. Explain what makes an element stab	le?
II. Answer the questions using the symbols	on the periodic table
1. What elements are in period 5?	
What element(s) is a	
2.halogens	6transition metals
3, lanthanides	7alkali metals
4alkaline earth metals	8.actinides
5noble gases	
What elements are in the sublevel	
9 s	11. d
10 p	12. f
What elements are	
13. metals,	
14. nonmetals	
15.metalloids,	
16. Which element is the most read	ctive nonmetal?
17. Which element is the most rea	ctive metal?
18.Which element below is the lea	ast reactive?
19. Which element has the largest	radius in period 4?
20.Which element has the largest	radius in the Alkali metals family?
21.Which element has the highest	ionization energy in period four?
22.Which elements has the second	l largest atomic number?
23. Which element has the largest	atomic mass?
24. Which element is located at po	riod 4, group 15?
25. Which element is located at per	eriod 5, group 2?
27 Which elements will form +1 ic	ons?
28. Which group of metals will hav	
29 Which group of nonmetals will	have the same properties
30. On the back, define all chapter:	5, section 3 vocabulary terms.

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i									
		100					z		



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**RJ Shavelson** 

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