

Answer Key

The Periodic Law

Section Quiz: History of the Periodic Table

As Marjorie provides, enter the name of the person whose work is described in each statement or select answer's best description.

1. In developing his periodic table, Mendeleev's observations were
 - a. periodic elements.
 - b. elements in liquid form.
 - c. isotopes.
 - d. **properties.**
2. Mendeleev's periodic table showed that all elements in each of the
 - a. column (groups) were similar in physical properties, such as atomic number.
 - b. **properties.**
 - c. periodic elements.
 - d. isotopes.
3. Mendeleev predicted that the gaps in his periodic table represented
 - a. ions.
 - b. unknown elements.
 - c. unstable isotopes.
 - d. **undiscovered elements.**
4. The periodic table, which today is periodic table based on increasing
 - a. atomic number, was
 - b. **Mendeleev.**
 - c. Moseley.
 - d. Rutherford.
5. The periodic law describes the physical and chemical properties of
 - a. **elements.**
 - b. isotopes.
 - c. ions.
 - d. compounds.

Periodic Law Section Quiz Answers

R Sanford

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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