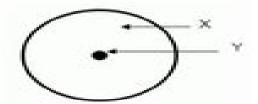
Test Periodic Table and Trends SCH3U Name:

Multiple Choice: Select the most correct answer (20K)

- 1. The mass number of an atom is determined by:
- a. the number of protons only
- b. adding the neutrons and protons
- c. adding the neutrons and electrons
- d. adding the protons and electrons



- 2. The letter "X" in the diagram above marks the:
- a. Region of greatest density
- b. Location of the protons
- c. Location of the neutrons
- d. Location of the electrons
- e. Nucleus
- 3. Most of the mass of the atom can be found in the:
- a. charges b. electron cloud c. electrons d. nucleus
- 4. Of the basic atomic particles, the one that would be attracted to a negatively charged metallic plate is the:
- a. proton b. neutron c. electron d. none of the above
- Isotopes (such as hydrogen-1, hydrogen-2, and hydrogen-3) are atoms of the same element that differ in:
- a, the number of protons in the nucleus
- b. the number of electrons
- c, the atomic number
- d, the number of neutrons in the nucleus
- e, the mass number
- Which of the following accurately summarizes the isotope Argon-40?
- a. 40 protons, 40 electrons, 18 neutrons
- b. 18 protons, 18 electrons, 22 neutrons
- 39.95 protons, 39.95 electrons, 21.05 neutrons
- d. 22 protons, 22 electrons, 18 neutrons

Periodic Table Test Answers

L Darling-Hammond

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