

Chapter 1: Limiting Reagents and Percent Yield (Pages 1-4)

Remember: Limiting reagent is the reactant that will completely run out first in a chemical reaction.

Excess: All reagents except the limiting reagent will be in excess. Excess is the amount of reagent that is left over after the reaction.

Remember: Limiting reagent is the reactant that will completely run out first in a chemical reaction.

Excess: All reagents except the limiting reagent will be in excess.



The reaction is balanced. The coefficients are 2, 1, and 2.

2H ₂	2 x 2 = 4 grams
O ₂	1 x 32 = 32 grams
2H ₂ O	2 x 18 = 36 grams
Total	4 + 32 = 36 grams

1. 4 grams of hydrogen and 32 grams of oxygen are mixed. How much water can be produced?



2. 4 grams of hydrogen and 32 grams of oxygen are mixed. How much water can be produced?

3. 4 grams of hydrogen and 32 grams of oxygen are mixed. How much water can be produced?



4. 4 grams of hydrogen and 32 grams of oxygen are mixed. How much water can be produced?



Percent Yield And Limiting Reactants Quiz

Ralph Thomas Myers



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