

Write Titrating Volume (Spreads: 1.00)

Name _____

Acetic Acid (pKa = 4.75) titrated with NaOH

a. 0.0500 M NaOH 0.0500 M Acetic Acid 10.00 mL

b. 0.0500 M NaOH 0.0500 M Acetic Acid 10.00 mL

c. 0.0500 M NaOH 0.0500 M Acetic Acid 10.00 mL

Acetic Acid (pKa = 4.75) titrated with NaOH

0.0500 M NaOH 0.0500 M Acetic Acid 10.00 mL

Species	Acid (pH, Base pKa, or Salt pKa)	Acids with M_1 (pKa = 4.75)	Conductivity (C or M)
H_2O	<u>7.0</u>	<u>1.0</u>	<u>1.0</u>
H_2O	<u>7.0</u>	<u>1.0</u>	<u>1.0</u>
H_2O	<u>7.0</u>	<u>1.0</u>	<u>1.0</u>
H_2O	<u>7.0</u>	<u>1.0</u>	<u>1.0</u>

Note Taking Guide 1101 Answers

Quentin Docter, Jon Buhagiar



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