

Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular $\text{C}_{10}\text{H}_{18}\text{O}_2$ (molecular)

Formula represents the number of _____ of _____
in a single molecule.

Empirical $\text{C}_{10}\text{H}_{18}\text{O}_2$ (molecular)

Formula represents the proportions of _____ of
positive and negative _____, hence formula
unit.

Atomic Mass

The proportions of the atoms proportions of _____
atoms together.

Masses

$\text{C}_{10}\text{H}_{18}\text{O}_2$ masses atoms

atoms are proportions of _____
atoms.

Subscripts

atoms atoms

atoms are proportions of _____

atoms

Note Taking Guide Episode 6chemical Formula Key

Rachel Sandford



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web oct 3 2022 as we mentioned in section 6 1 exponential and logarithmic functions are used to model a wide variety of behaviors in the real world in the examples that follow

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