

- 10. The reaction is exothermic because the products have a lower energy than the reactants.
- 11. The activation energy is the energy difference between the reactants and the products.
- 12. The reaction is exothermic because the products have a lower energy than the reactants.
- 13. The reaction is exothermic because the products have a lower energy than the reactants.



Figure 1



Figure 2

Figure 1 shows the potential energy diagram for the reaction of hydrogen and oxygen to form water. Figure 2 shows the potential energy diagram for the reaction of hydrogen and oxygen to form hydrogen peroxide.

Answer the questions in the margin.

Sca2hp Science Paper Unit January

D Keegan



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