

Periodic Trends Worksheet

Directions: Use your notes to answer the following questions.

1. Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium.

Oxygen < Carbon < Aluminum < Potassium

2. Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.

Neon < Aluminum < Sulfur < Oxygen

3. Why does fluorine have a higher ionization energy than iodine?

Fluorine has nine protons attracting 9 electrons which are much closer to the nucleus compared to iodine. This results in smaller size of Fluorine than iodine making it difficult to remove an electron from the atom.

4. Why do elements in the same family generally have similar properties?

Because they have same number of electrons in the outer shell (valence electrons) which take part in chemical reaction.

5. Indicate whether the following properties increase or decrease from left to right across the periodic table.

- | | |
|--|------------------|
| a. atomic radius (excluding noble gases) | Decreases |
| b. first ionization energy | Increases |
| c. electronegativity | Increases |

6. What trend in atomic radius occurs down a group on the periodic table? What causes this trend?

Atomic radius **increases** down the group on the periodic table. As we compare the elements down the group, the effective nuclear charge increases, but at the same time the outermost electrons are found in the shell that is farther away from the nucleus. Also the number of screening electrons increases. This results in reduced attraction between the nucleus and the outermost electrons.

7. What trend in ionization energy occurs across a period on the periodic table? What causes this trend?

Ionization energy **increases** from left to right across the period. The atomic size becomes smaller from left to right. So it becomes harder to remove electron from the atom. Hence the energy required to do so (ionization Energy) increases.

Periodic Trends Worksheet Extension Questions Key

**Robert C. Smoot, Smoot, Richard G.
Smith, Jack Price**



Periodic Trends Worksheet Extension Questions Key:

Unveiling the Energy of Verbal Art: An Emotional Sojourn through **Periodic Trends Worksheet Extension Questions Key**

In a global inundated with screens and the cacophony of instantaneous transmission, the profound energy and mental resonance of verbal art usually fade in to obscurity, eclipsed by the regular barrage of noise and distractions. However, located within the lyrical pages of **Periodic Trends Worksheet Extension Questions Key**, a captivating work of fictional beauty that pulses with fresh feelings, lies an unique journey waiting to be embarked upon. Published by way of a virtuoso wordsmith, this exciting opus courses readers on an emotional odyssey, softly exposing the latent potential and profound influence stuck within the complex web of language. Within the heart-wrenching expanse with this evocative evaluation, we shall embark upon an introspective exploration of the book is key styles, dissect their charming publishing style, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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