

Periodic Table of the Elements

	+1	+2											+3	+4	+5	+6	+7	0												
1	H 1.008	He 4.003											B 10.81	C 12.01	N 14.01	O 16.00	F 18.99	Ne 20.18												
2	Li 6.941	Be 9.012											Al 26.98	Si 28.09	P 30.97	S 32.07	Cl 35.45	Ar 39.95												
3	Na 22.99	Mg 24.31	Sc 44.96	Ti 47.88	V 50.94	Cr 52.00	Mn 54.94	Fe 55.85	Co 58.93	Ni 58.69	Cu 63.55	Zn 65.38	Ga 69.72	Ge 72.64	As 74.92	Se 78.96	Br 79.90	Kr 83.80												
4	K 39.10	Ca 40.08	Sc 44.96	Ti 47.88	V 50.94	Cr 52.00	Mn 54.94	Fe 55.85	Co 58.93	Ni 58.69	Cu 63.55	Zn 65.38	Ga 69.72	Ge 72.64	As 74.92	Se 78.96	Br 79.90	Kr 83.80												
5	Rb 85.47	Sr 87.62	Y 88.91	Zr 91.22	Nb 92.91	Mo 95.94	Tc 98.91	Ru 101.07	Rh 102.91	Pd 106.42	Ag 107.87	Cd 112.41	In 114.82	Sn 118.71	Sb 121.76	Te 127.60	I 126.91	Xe 131.29												
6	Cs 132.91	Ba 137.33	La 138.91	Ce 140.12	Pr 140.91	Nd 144.24	Pm 144.91	Sm 150.36	Eu 151.96	Gd 157.25	Tb 158.93	Dy 162.50	Ho 164.93	Er 167.26	Tm 168.93	Yb 173.05	Lu 174.97													
7	Fr [223]	Ra [226]	Ac [227]	Th [232]	Pa [231]	U [238]	Np [237]	Pu [244]	Am [243]	Cm [247]	Bk [247]	Cf [251]	Es [252]	Fm [257]	Mn [288]	Ni [289]	Pu [289]	Am [287]	Cm [284]	Bk [282]	Cf [281]	Es [277]	Fm [276]	Mn [270]	Ni [269]	Pu [269]	Am [267]	Cm [265]	Bk [264]	Cf [262]
8	Ce [140.12]	Pr [140.91]	Nd [144.24]	Pm [144.91]	Sm [150.36]	Eu [151.96]	Gd [157.25]	Tb [158.93]	Dy [162.50]	Ho [164.93]	Er [167.26]	Tm [168.93]	Yb [173.05]	Lu [174.97]																
9	Th [232]	Pa [231]	Np [237]	Pu [244]	Am [243]	Cm [247]	Bk [247]	Cf [251]	Es [252]	Fm [257]	Mn [288]	Ni [289]	Pu [289]	Am [287]	Cm [284]	Bk [282]	Cf [281]	Es [277]	Fm [276]	Mn [270]	Ni [269]	Pu [269]	Am [267]	Cm [265]	Bk [264]	Cf [262]				

The only metal that is liquid at room temperature.

Period 3
element with
7 valence
electrons.

Semi-conductor
metalloid used
in electronics.

Only one valence electron, but not part of a family.

Extremely reactive non-metal.

Group that will form 2+ cations.

The only non-metal that is liquid at room temperature.

Completely
un-reactive family
with complete
valence shell.

Extremely reactive metal.

Group with 4 valence electrons.

Periodic Trends Review

Lei Huang

A red circular graphic with a gradient, appearing as a semi-circle or a stylized arrow pointing to the right, located on the right side of the slide.

Periodic Trends Review:

Unveiling the Power of Verbal Beauty: An Mental Sojourn through **Periodic Trends Review**

In a world inundated with screens and the cacophony of quick conversation, the profound energy and mental resonance of verbal artistry usually diminish in to obscurity, eclipsed by the continuous assault of noise and distractions. However, set within the musical pages of **Periodic Trends Review**, a interesting function of fictional beauty that pulses with natural feelings, lies an unforgettable journey waiting to be embarked upon. Penned by a virtuoso wordsmith, this mesmerizing opus books readers on a mental odyssey, lightly exposing the latent potential and profound affect stuck within the intricate internet of language. Within the heart-wrenching expanse of this evocative analysis, we shall embark upon an introspective exploration of the book is main styles, dissect its interesting writing design, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

<https://crm.avenza.com/results/Resources/HomePages/Older%20Gmade%20Scoring%20Manual.pdf>

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