

Periodic Trends Guided Note Sheet

Name _____

Period _____

Date _____

Fill in the blanks. Some blanks have word choices written after them to help you. Refer to the drawings and questions from the first page to help you with the fill in the other pages.

I. Atomic Radius in a Group

- A. As you move down a group, the nucleus becomes _____ (stronger, weaker) because more _____ (electrons, protons) are being added to it. This stronger nucleus can pull the electrons in more tightly, making the atoms **smaller; smaller**.
- B. _____ (Protons, Electrons) are also added to the atom but are added to _____ (larger, smaller, the same) energy levels resulting in **larger** atoms.
- C. The increased number of inner electrons shield (or block) the _____ (positive, negative) pull of the nucleus.
- D. So, electron shielding _____ (increases, decreases) causing the electrons on the outermost energy levels to be even further from the nucleus than expected.
- E. This causes the atoms at the bottom of a Group to be even _____ (larger, smaller) than expected.
- F. Therefore, atomic radius _____ (increases, decreases) as you move down a group.

II. Atomic Radius in a Period

- A. As you move right across a Period, the nucleus becomes _____ (stronger, weaker) because more _____ (electrons, protons) are being added to it.
- B. _____ (Protons, Electrons) are also added but they are added to _____ (the same, a larger, a smaller) energy level.
- C. This causes the amount of electron shielding from electrons on inner energy levels to remain _____ (constant, varied).
- D. Therefore, electrons are pulled in _____ (more, less) tightly because they feel the pull from nucleus more strongly.
- E. This causes the atoms on the right of the Periodic Table to be _____ (larger, smaller) than those on the left.
- F. So, atomic radius _____ (increases, decreases) as you move across a Period from left to right.

III. Sizes of an Ion Versus Sizes of an Atom (Digital Textbook—Section 11.2)

- A. Metal cations are _____ (smaller, larger) than the neutral atoms.
- B. because _____
- C. Nonmetal anions are _____ (smaller, larger) than the neutral atoms.
- D. because _____

IV. Ionization Energy in a Group

- A. Ionization energy is the energy required to _____ the outermost electron from an atom.
- B. As you move down a Group, the nucleus becomes _____ (stronger, weaker) because there are _____ (more, fewer) protons in it.
- C. **Constant**, as you move down a group, there are _____ (fewer, more) energy levels making the atom _____ (larger, smaller).
- D. In atoms at the top of a group, the outermost electrons are closer to the nucleus so the electrons can feel the pull of nucleus _____ (more, less) strongly.
- E. The outermost electrons also feel the pull of the nucleus less strongly because there is _____ (less, more) electron shielding from the _____ (decreased, increased) number of inner electrons.
- F. Therefore, _____ (more, less) energy is required to remove the outermost electron from a larger atom.

Periodic Trends Review Notes Sheet Answers

Puja Mehta



Periodic Trends Review Notes Sheet Answers:

Manual for performance evaluation of regulatory authorities seeking designation as WHO-listed authorities

World Health Organization, 2023-11-14 This manual provides operational and technical details for the performance evaluation PE exercise that must be conducted for a regulatory authority RA to achieve listing as a WHO listed authority WLA in relation to each regulatory function The PE manual should be read in conjunction with the Operational guidance for evaluating and publicly designating regulatory authorities as WHO listed authorities The Operational Guidance For the purposes of this document the term regulatory authority RA unless otherwise stated may refer to either a national regulatory authority NRA or a regional regulatory system RRS The basis for designation as a WLA is provided by the Global Benchmarking Tool GBT which is complemented by a series of PE activities designed to establish a detailed picture of how the regulatory system performs on relevant regulatory processes including how consistently it adheres to quality procedures and how well it delivers the desired regulatory outputs in accordance with good regulatory practices *Wiley CPA Examination Review, Problems and Solutions* Patrick R. Delaney, O. Ray Whittington, 2012-05-23 The 1 CPA exam review self study leader The CPA exam review self study program more CPA candidates turn to take the test and pass it Wiley CPA Exam Review 39th Edition contains more than 4 200 multiple choice questions and includes complete information on the Task Based Simulations Published annually this comprehensive two volume paperback set provides all the information candidates need to master in order to pass the new Uniform CPA Examination format Features multiple choice questions new AICPA Task Based Simulations and written communication questions all based on the new CBT e format Covers all requirements and divides the exam into 47 self contained modules for flexible study Offers nearly three times as many examples as other CPA exam study guides With timely and up to the minute coverage Wiley CPA Exam Review 39th Edition covers all requirements for the CPA Exam giving the candidate maximum flexibility in planning their course of study and success Knowledge and Technology Management in Virtual Organizations: Issues, Trends, Opportunities and Solutions Putnik, Goran D., Cruz-Cunha, Maria

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