

Periodic Trends Worksheet

Name: _____ Grade: _____ ID: _____

Directions: Use your notes to answer the following questions.

- 1 Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium.

Ans: O, C, Al, K

- 2 Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.

Ans: Ne, Al, S, O

- 3 Why does fluorine have a higher ionization energy than iodine?

Ans: Fluorine has a smaller atomic radius, so the proton can exert a greater pull on electron (Valence shell is close)

- 4 Why do elements in the same family generally have similar properties?

Ans: They have the same valence shell electron arrangement.

- 5 Indicate whether the following properties increase or decrease from left to right across the periodic table.

- a. atomic radius (excluding noble gases) **Decrease**
- b. first ionization energy **Increase**
- c. electronegativity **Increase**

- 6 What trend in atomic radius occurs across the periodic table? What causes this trend?

Ans: Atomic radius increases down a group because energy levels (shells) are added.

- 7 What trend in ionization energy occurs across a period on the periodic table? What causes this trend?

Ans: Ionization energy increases across a period because as elements become less metallic, it requires more energy to remove an electron.

- 8 Circle the atom in each pair that has the largest radius.

a. Al or B

d. O or F

b. Na or Al

e. Br or Cl

c. S or O

f. Mg or Ca

Periodic Table Trends Answer Key

Raffaela Di Napoli

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Periodic Table Trends Answer Key:

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Constructing Measures Mark Wilson, 2004-12-13 Constructing Measures introduces a way to understand the advantages and disadvantages of measurement instruments how to use such instruments and how to apply these methods to develop new instruments or adapt old ones The book is organized around the steps taken while constructing an instrument It opens with a summary of the constructive steps involved Each step is then expanded on in the next four chapters These chapters develop the building blocks that make up an instrument the construct map the design plan for the items the outcome space and the statistical measurement model The next three chapters focus on quality control They rely heavily on the calibrated construct map and review how to check if scores are operating consistently and how to evaluate the reliability and validity evidence The book introduces a variety of item formats including multiple choice open ended and performance items projects portfolios Likert and Guttman items behavioral observations and interview protocols Each chapter includes an overview of the key concepts related resources for further investigation and exercises and activities Some chapters feature appendices

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Hands-On General Science

Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life

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