
The Periodic Law

Section Quiz: History of the Periodic Table

As Marjorie provided, enter the name of the person whose work is described in each statement or select answer's best description.

1. In developing his periodic table, Mendeleev's observations were
 - a. periodic elements.
 - b. elements in liquid form.
 - c. isotopes.
 - d. **properties.**
2. Mendeleev's periodic table showed all elements in order of
 - a. increasing atomic mass because he grouped together elements with similar
 - b. **properties.**
 - c. atomic numbers.
 - d. isotopes.
3. Mendeleev predicted that the gaps in his periodic table represented
 - a. ions.
 - b. unknown elements.
 - c. unstable isotopes.
 - d. **undiscovered elements.**
4. The periodic table came before a periodic table based on increasing
 - a. atomic number was
 - b. **isotopes.**
 - c. Mendeleev.
 - d. classification.
5. The periodic law describes the physical and chemical properties of
 - a. **elements.**
 - b. isotopes.
 - c. ions.
 - d. compounds.

Periodic Law Section Quiz Answers

Terry C. Jones



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