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Arrhenius bases: an Arrhenius base is defined as any species that increases the concentration of hydroxide ions (OH^-) in aqueous solution. An example of an Arrhenius base is the highly soluble sodium hydroxide (NaOH). Sodium hydroxide dissociates in water as follows:

$$\text{NaOH} \rightarrow \text{Na}^+ + \text{OH}^-$$

In water, sodium hydroxide

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