

Chapter 7: Chemical Reaction Guided Notes

I. Chemical Changes in Matter

A. Chemical Reaction

1. A change in which one or more substances are converted to _____ substances.

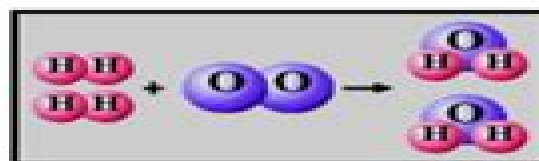


B. Law of Conservation of Matter

1. In a chemical reaction, matter is not _____ or _____.
2. Atoms can only be _____.
3. Discovered by Lavoisier.

_____ Hydrogen

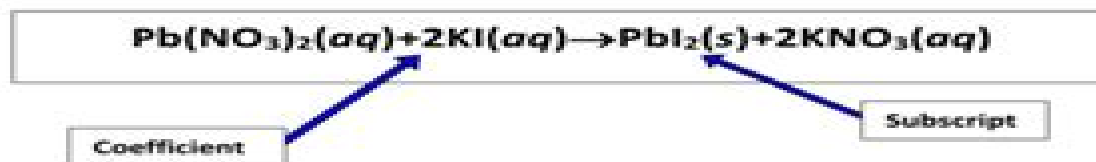
_____ Oxygen



_____ Hydrogen

_____ Oxygen

C. Chemical Equations



1. _____ - # of units of each substance.
2. Individual atom = **atom**
 $2\text{Mg} \Rightarrow 2$ _____ of magnesium
3. Covalent substance = **molecule**
 $3\text{CO}_2 \Rightarrow 3$ _____ of carbon dioxide
4. Ionic substance = **unit**
 $4\text{MgO} \Rightarrow 4$ _____ of magnesium oxide

SYMBOL	MEANING
$\longrightarrow$	produces, forms
+	plus, and
(s)	solid
(l)	liquid
(g)	gas
(aq)	aqueous (solid dissolved in water)
Δ	the reactants are heated

Notetaking Guide Answer Keychemical Reactions

Chao Zhang



Notetaking Guide Answer Keychemical Reactions:

Decoding **Notetaking Guide Answer Keychemical Reactions**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Notetaking Guide Answer Keychemical Reactions**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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