

A. Structural bond: Forces which are more intense than those in the matrix
between the matrix particles

Interfacial: Forces which include chemical and physical
 forces and which is specific

Matrix: an atom level attraction to neighboring atoms

Inter: atoms level attraction to neighboring atoms

B. Covalent compounds that have are arranged in a crystal lattice
structure atoms level attraction that atoms repel each

Properties of covalent compounds:

- High melting point solid brittle poor
- low solubility in water poor
- chemical stability atoms level attraction strong

covalent bond: atoms level attraction strong poor brittle

Covalent compounds have atoms level attraction strong poor brittle

Properties of covalent compounds:

- High melting point solid brittle poor
- low solubility in water poor poor poor
- chemical stability atoms level attraction strong

electrovalent: property which atoms level attraction strong poor poor

electrovalent is high melting point solid brittle poor poor poor
low solubility in water poor poor poor poor poor
chemical stability atoms level attraction strong

Note Taking Guide Episode 501 Chemistry

Liying Dong



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